



4th WRC
SYMPOSIUM

INNOVATION IN EVERY DROP 11 TO 13 SEPTEMBER



FUTURE
WATER

SYSTEM-WIDE MODELLING FOR FUTURE WASTE AND RESOURCE RECOVERY FACILITIES

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~~WASTEWATER TREATMENT~~
FOR EFFLUENT QUALITY



WATER AND RESOURCE
RECOVERY – OPTIMISED
GENERATION OF
PRODUCTS
(UTILISATION OF MODELS
FOR PREDICTED SYSTEM
PERFORMANCE)

Performance
Data



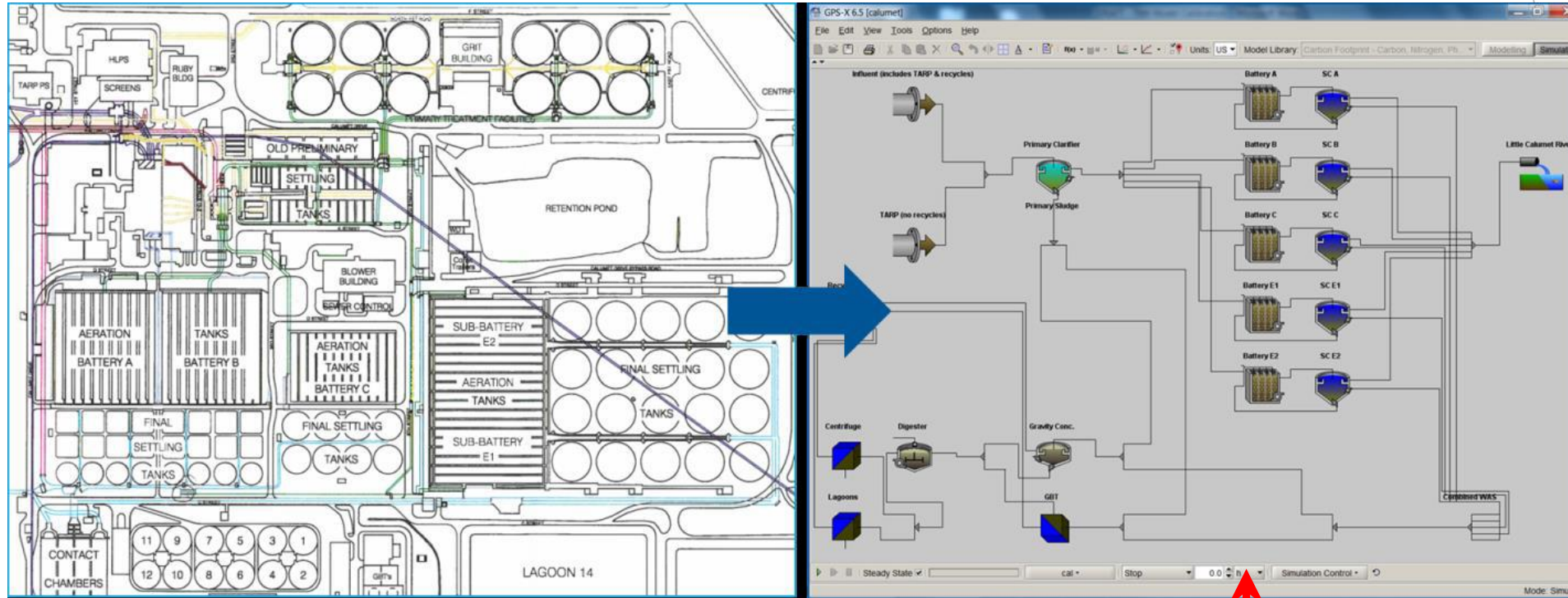
Virtual Production



Real Production



Modelling Socially, Economically and Environmentally Sustainable Waste Treatment Systems of the Future:



- **Include, Evaluative Performance Indices for Effluent Quality (EQI) and Cost (OCI) at System-Wide Level.**

Recovered Products (i.e., Nitrogen, Methane, Organic fertiliser, Phosphorus, Water, etc).

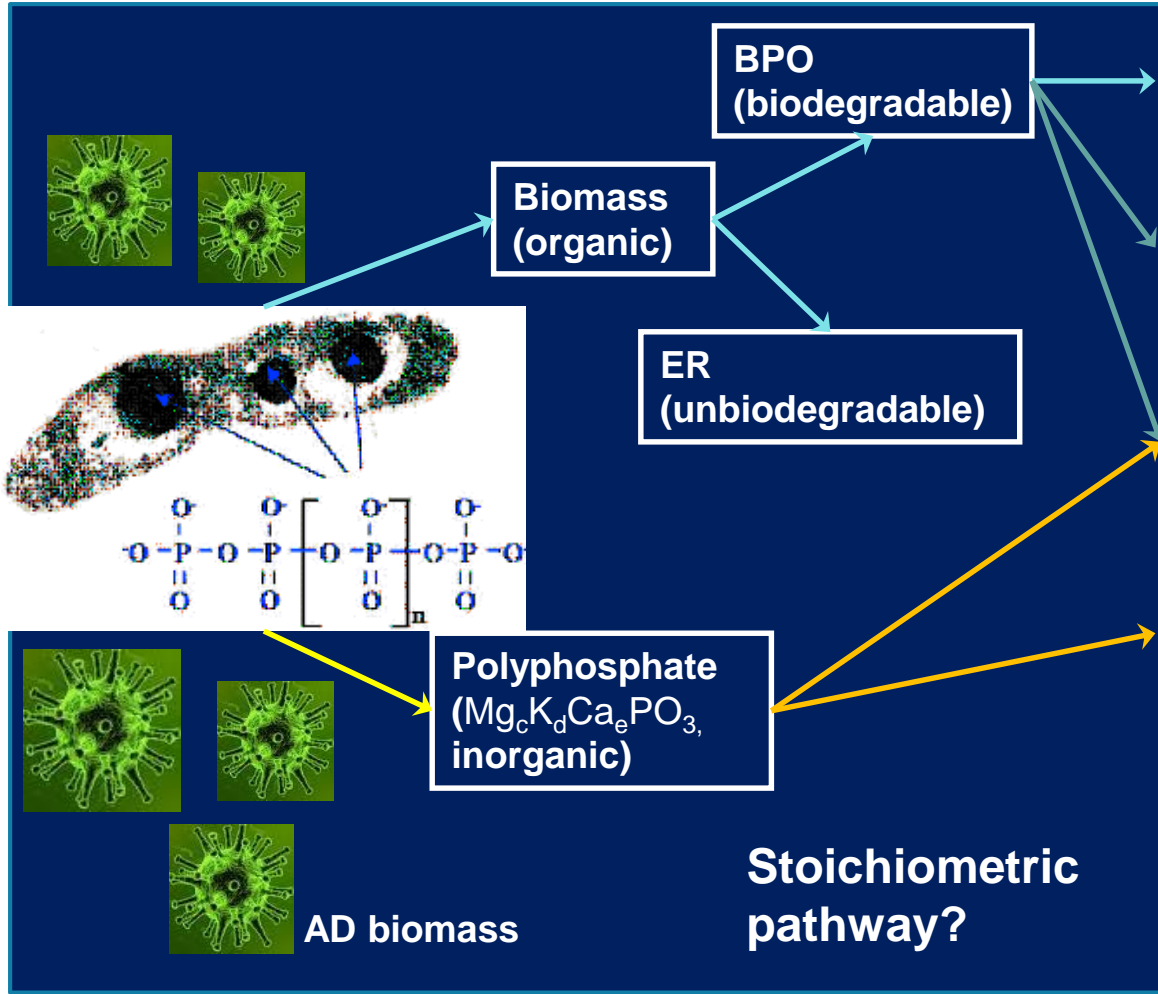
Generate experimental data that can answer questions raised from modelling of phosphorus (P) removal systems.

Requires:

- PAO behaviour (mainly anaerobic polyphosphate release)
- Chemically dosed components

UCT plant wide dynamic model (PWM_SA) be utilised towards transforming WWTPs to WRRFs

- Review fate of WWTP products
- Develop evaluative protocol for entire system (including fate of products)
- Case Studies on evaluation of full scale systems



Biogas (CH₄ & CO₂)

Ammonia (NH₄⁺)

Phosphates (H₂PO₄⁻, HPO₄²⁻, etc.)

Metals (i.e., Mg, K, Ca, etc.)

Mineral Precipitates, i.e. Struvite, ACP, etc

Anaerobic Digester System Products

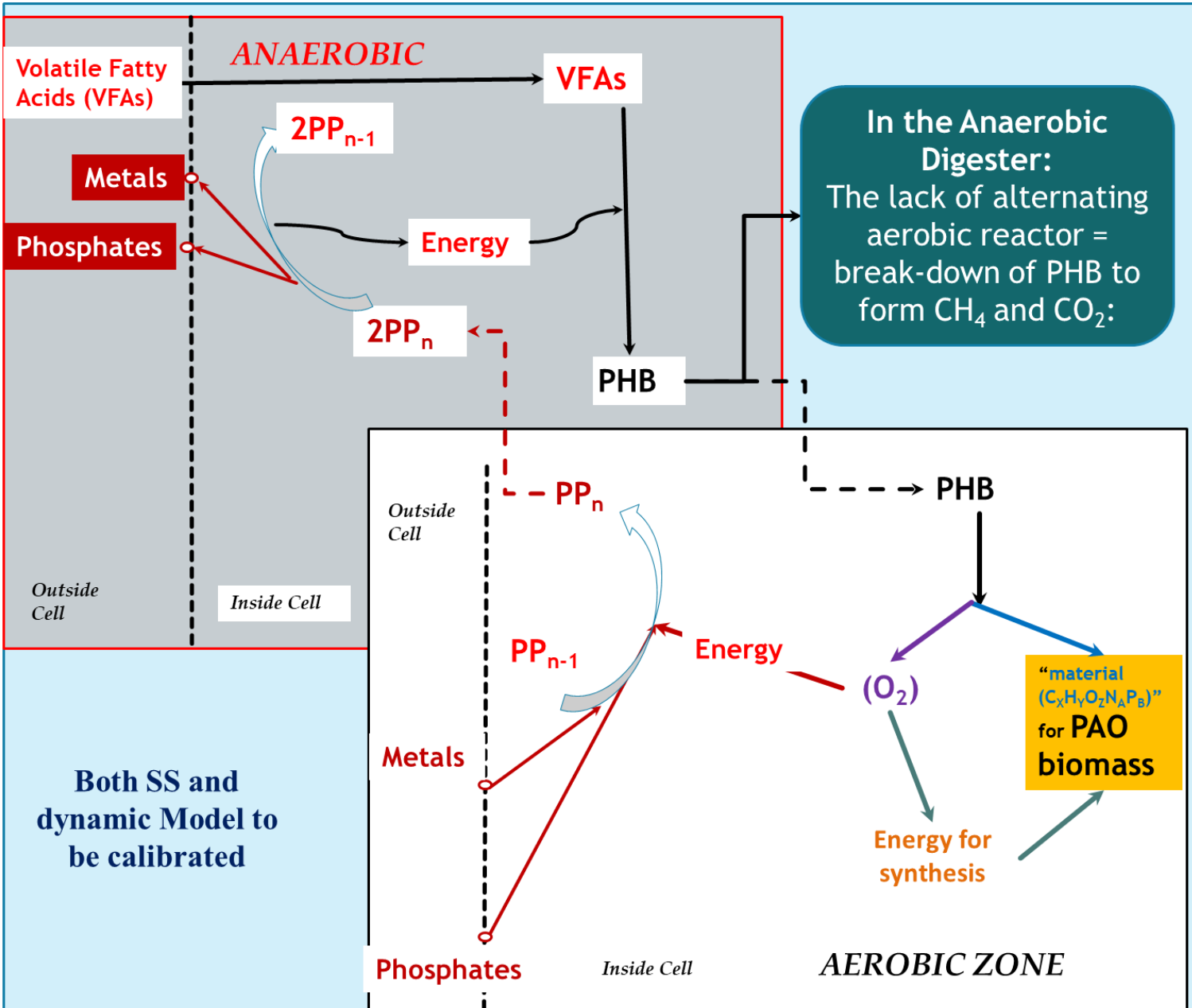


MgNH₄PO₄

Ca₃(PO₄)₂

CaCO₃

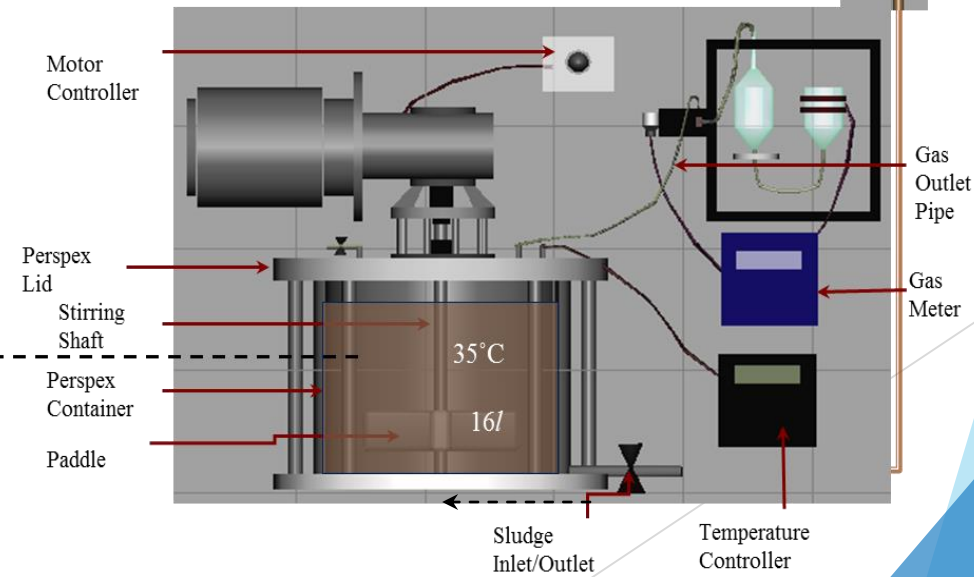
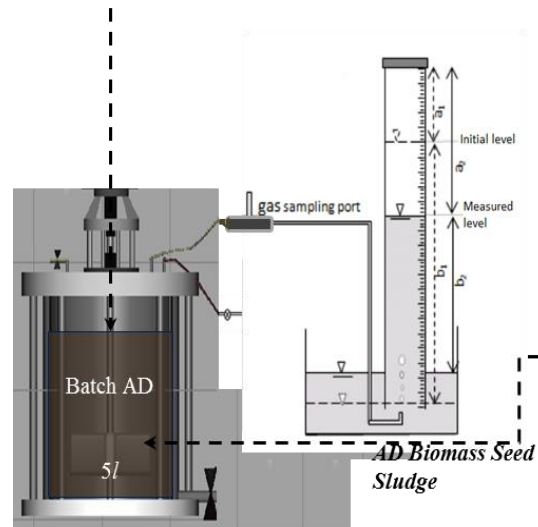
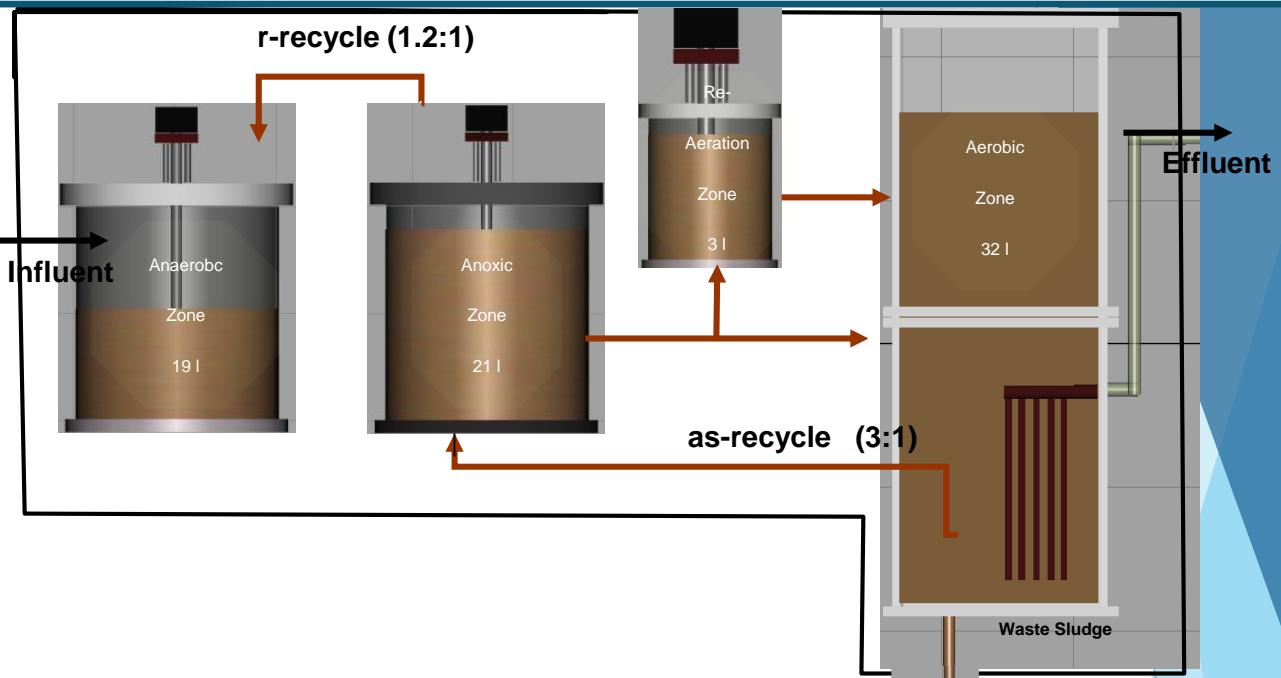
PAO BEHAVIOR IN AS AND AD SYSTEMS



In the Anaerobic Digester:
 The lack of alternating aerobic reactor = break-down of PHB to form CH₄ and CO₂:

EXPERIMENTS TO MODEL P REMOVAL

The AD system fed enhanced culture of PAO's to (1) Identify correct AD stoichiometric Pathways for PAOs; (2) Measure PAO ER fraction; (3) Determine Kinetics of PAO breakdown in AD (4) Generate data to Validate AD model.



Research Questions

- Are Mathematical models that include biological and chemical P removal sufficient to inform optimal design?
- Should chemical P removal using ferric and/or Aluminium ions be used to supplement biological P removal in activated sludge system?
- Is it cost effective and efficient to remove P biologically without chemical addition (ferric and/or Aluminium ions) in activated sludge system?
- What is the fate of the resulting chemicals binding phosphorus in the waste activated sludge?

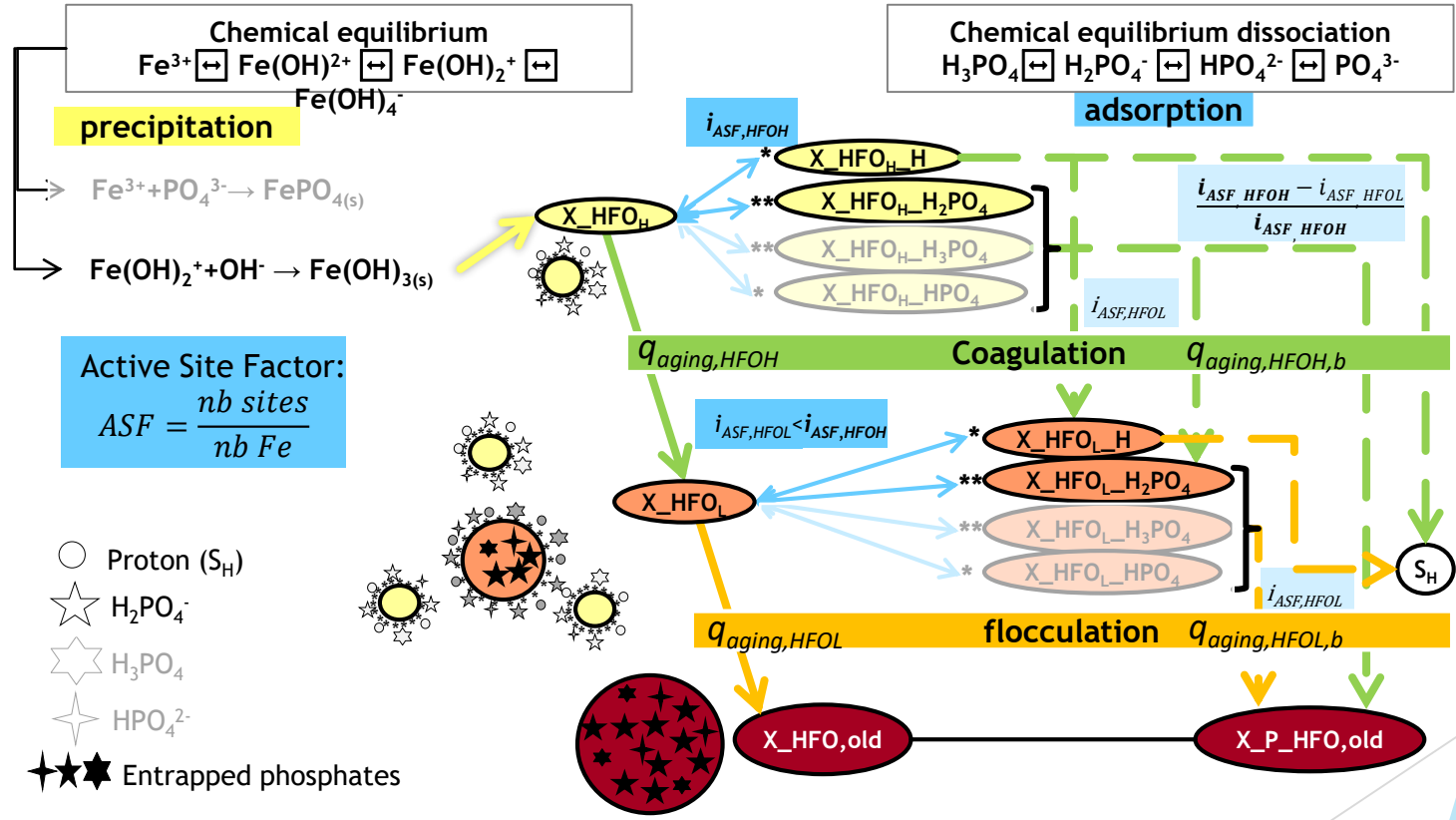
Research Approach:

- Extend Model and utilize data from Literature to explore the chemical P removal via pre-precipitation, co-precipitation and post-precipitation
- Experiments used where research gaps identified (i.e., data unavailable)

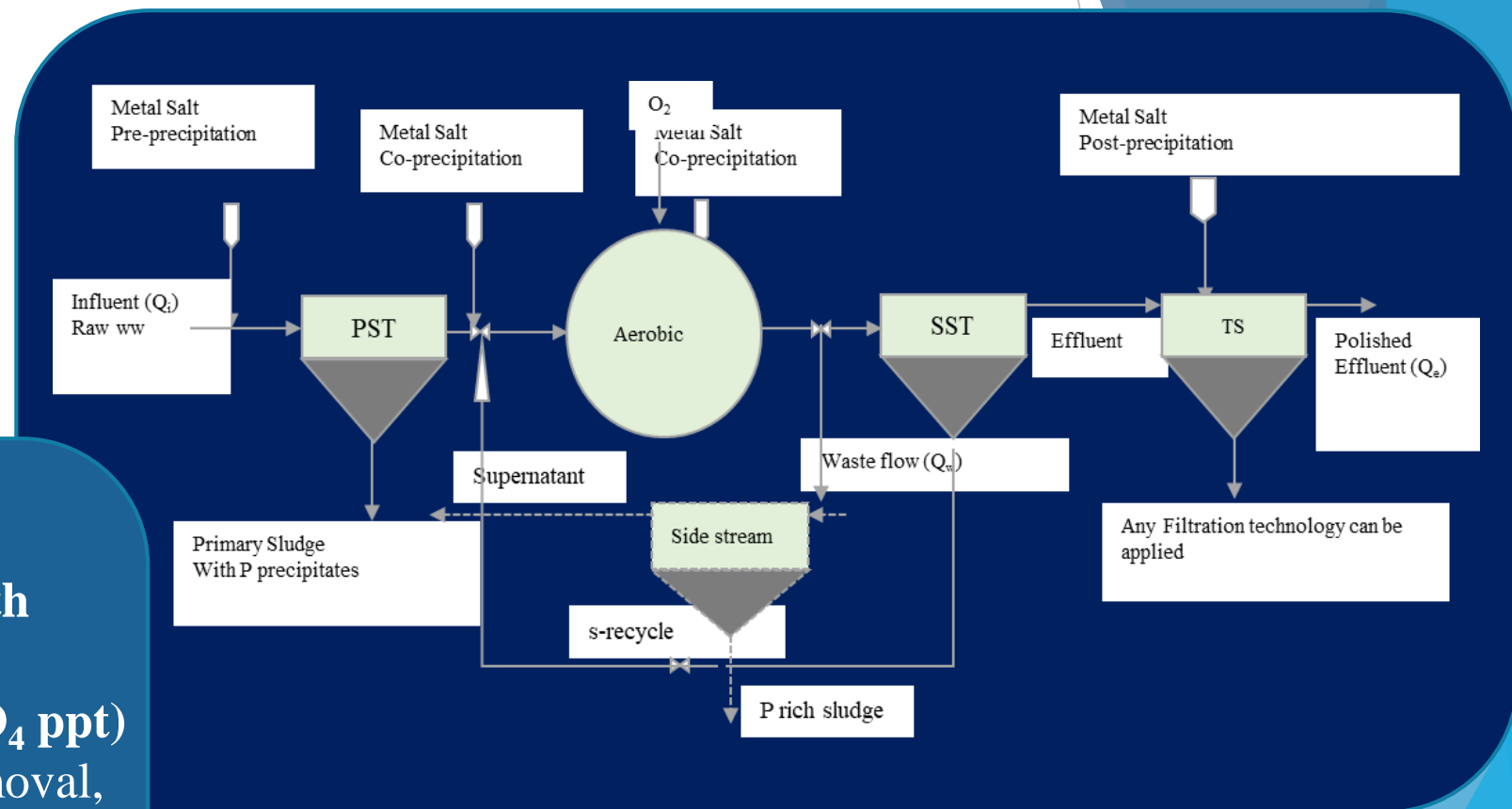
Hauduc et al., 2015

- 2/1 Precipitation Processes
- 8/4 Adsorption Processes
- 4 Aging Processes
- 12/8 State Variables

Simplified model

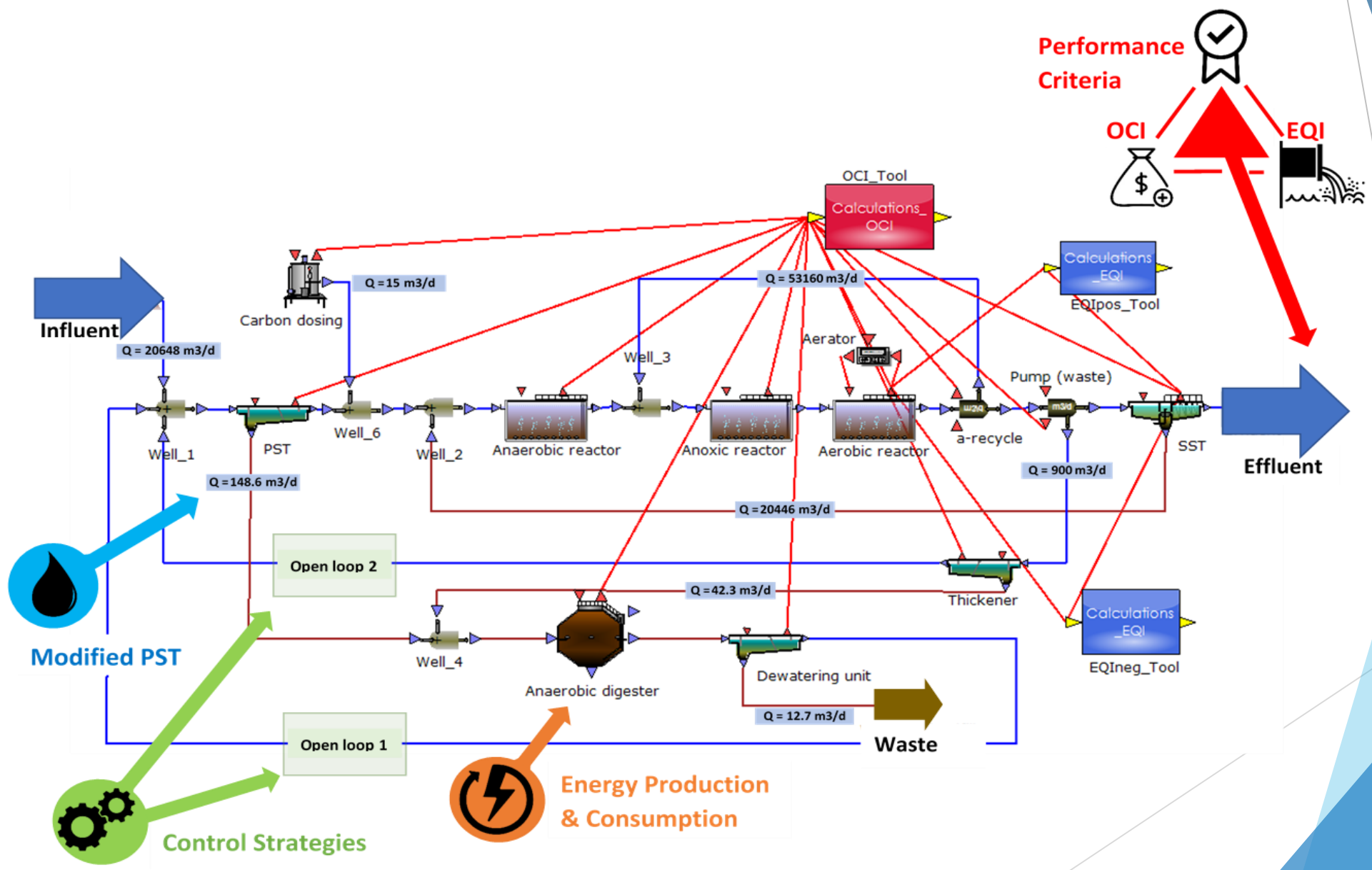


Extension of PWMSA model to include above chemical precipitation model



- Calibration of the model (E.M.) with data from (De Haas; FeCl co precipitation), (Power, 1991; $AlSO_4$ ppt)
- Evaluate model predictions of P removal, when done (i) biologically, (ii) chemically for co-precipitation, and (iii) when (i) and (ii) are combined
- Assess impact of pre-precipitation

Developing an Evaluative Protocol



$$EQI_{water} = \frac{1}{T \cdot 1000} \int_{t_0}^{t_{end}} (\beta_{TSS} \cdot TSS(t) + \beta_{COD} \cdot COD(t) + \beta_{FSA} \cdot FSA(t) + \beta_{NO} \cdot NO(t) + \beta_{OP} \cdot OP(t) + \beta_x \cdot X(t)) \cdot Q_e(t) \cdot dt$$

Eq. 1

- X(t): Any other pollutant considered in future evaluations
- Flexibility in the weighting factors

$$EQI_{gas} = \frac{1}{T \cdot 1000} \int_{t_0}^{t_{end}} (\beta_{CO_2} \cdot FCO_2(t) + \beta_{CH_4} \cdot FCH_4(t) + \beta_{N_2O} \cdot FN_2O(t)) \cdot dt$$

Eq. 2

T : Total length of evaluation period (days)

β : Pollutant weighting factor ($\beta_{CO_2} = 1; \beta_{CH_4} = 32; \beta_{N_2O} = 281$)

F : Flux of gas evolved (kg/d)

Works in Progress: Formulation for EQI Sludge?

OCI

$$= (AE + PE - MP + ME + HE) \cdot \text{Energy cost} + SP \cdot \text{Sludge disposal cost} + EC$$

$$\cdot \text{Carbon cost} + \text{Metals dosed} \cdot \text{Metal cost} + \text{Lime dosed} \cdot \text{Lime cost} - NR$$

$$\cdot \text{Market price} + \text{Fines}$$

Eq. 3

AE	: Aeration energy (kWh/d)
PE	: Pumping energy (kWh/d)
SP	: Sludge produced (kgTSS/d)
EC	: External carbon addition (kgCOD/d)
ME	: Mixing energy (kWh/d)
MP	: Energy from methane produced (kWh/d)
HE	: Total heat energy required by anaerobic digester for sludge treatment (kWh/d)
NR	: Nutrient recovered; e.g Struvite (kg/d)

Delivering a new generation of engineering tools to simulate resource recovery options and to evaluate these novel technologies in line with the circular economy paradigm



Development and validation of engineering tools for resource recovery technologies for South African treatment plants



Interfacing engineering tools within system-wide modelling approaches



Demonstrating novel process simulation and scenario analysis options using full-scale case studies



Promote the development of a WRRF mathematical modelling community



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